

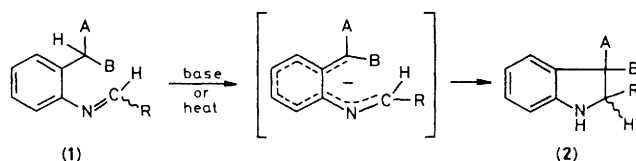
Enantioselective 1,5-Electrocyclization. Synthesis of Optically Active Indolines

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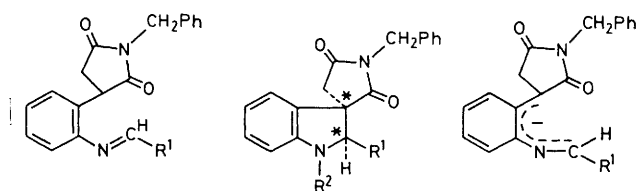
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Intramolecular base-catalysed 1,5-electrocyclization of aldehyde imines derived from *o*-aminophenyl succinimide (3) in the presence of a chiral alcohol yields chiral indolines (4) with optical purities ranging from 17 to 31%.

Intermolecular enantioselective additions of nucleophiles to electrophilic alkenes sometimes give impressive results in terms of optical yields.¹ The corresponding intramolecular variant, although only occasionally studied, has also been shown to afford appreciable asymmetric induction.² Recently a novel intramolecular addition of a carbanionic species on to an imine was reported³ which could be represented in a general way by the reaction (1) → (2) (Scheme 1). While enantioselective intermolecular imine additions have been studied before⁴ the reaction which we report now differs essentially in mechanism and is termed more accurately as a 1,5-electrocyclization.⁵ For the latter type of reaction no examples of asymmetric induction have been reported so far.



Scheme 1



(3) a, R¹ = Bu^t
b, R¹ = Prⁿ
c, R¹ = Ph
d, R¹ = CH=CHPh

(4) R² = H
(6) R² = Ac

(5)

The intramolecular imine cyclization is exemplified by the base-catalysed conversion of the imine (3a) into the indoline (4a) which proceeds in a stereocontrolled fashion in a bulky alcoholic solvent. The compounds (4) are of value in a short and general synthesis of indole alkaloids.⁶ Since the origin of the stereoselective effect was assumed to reside in a particularly strong association between the hydroxy-moiety of the alcohol and the reactive intermediate,⁷ most probably the anion (5), the use of a chiral alcohol could profoundly influence the stereochemical outcome. We now report that addition of a chiral co-solvent in the 1,5-electrocyclization of (3a-d) leads to the optically active indolines (4a-d), thereby in principle allowing the enantioselective synthesis of Aspidosperma alkaloids.

In a typical reaction (–)-menthol {0.500 g; [α]_D²⁰ (room temp.) –42°, c = 6.3, CHCl₃} was dissolved in tetrahydrofuran (THF) (10 ml) and BuⁿLi (0.5 mmol; 20% soln. in hexane) followed by a solution of (3a) (0.5 mmol) in toluene (1 ml) added. The mixture was stirred for 3 h at 0 °C, and the menthol was removed by column chromatography on silica gel with EtOAc–cyclohexane (1:10) as eluant affording chemically pure (4a) (139 mg; 62%) as an oil {[α]_D²⁰ (room temp.) –55°, c = 1.0, CHCl₃} structurally identical with the product, m.p. 159–160 °C, formed upon treatment of (3a) with NaOBu^t. The enantiomeric purity of (4a) was determined via 250 MHz ¹H n.m.r. analysis of the benzylic AB pattern using the shift reagent Eu(hfc)₃,† and this indicated an enantiomeric excess (e.e.) value of 31% (±2). Upon fractional crystallization of (4a) from EtOAc–hexane the pure enantiomer was obtained; m.p. 137–138 °C {[α]_D²⁰ (room temp.) –152°, c = 0.92, CHCl₃}. Since the type of cation and/or the reaction

† Eu(hfc)₃: tris-[3-(heptafluoropropylhydroxymethylene)-(–)-camphorato]europium(III), >99%, Gold Label, Aldrich.

Table 1. Enantioselective synthesis of the indolines (4) from the imines (3).

Entry	R ¹	Alcohol	T/°C	Cation	Product	t/min	e.e. (±2%) of (4)	Chemical yield (%)	[α] ₅₇₈ (room temp.) (CHCl ₃)
1	Bu ^t	(-)-Menthol	0	Li	(4a)	180	31 ^a	62	-55° (c 1.0)
2	"	(-)-Menthol	Room temp.	Li	(4a)	90	27 ^b	65	-44° (c 1.7)
3	"	(-)-Menthol	"	Na	(4a)	90	28 ^b	52	-43° (c 9.3)
4	"	(-)-Borneol	0	Li	(4a)	180	17 ^b	73	-25° (c 4.0)
5	Pr ⁿ	(-)-Borneol	0	Li	(4b)	15	18 ^a	70	-10° (c 3.9)
6	"	(-)-Menthol	0	Li	(4b)	15	18 ^b	40	-10° (c 13.9)
7	Ph	(-)-Menthol	-40	Li	(6c)	120	29 ^a	64	+4.7 (c 14.0)
8	PhCH=CH ₂ (E)	(-)-Menthol	0	Li	(6d)	15	22 ^a	85	+44° (c 7.0)

^a e.e. determined by ¹H n.m.r. spectroscopy with Eu(hfc)₃, ^b e.e. calculated from [α]₅₇₈ (room temp.).

Table 2. C.d. data for (4a), (6d), and (7).

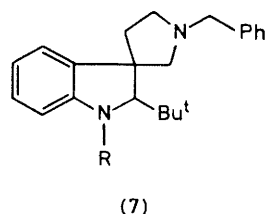
Compound	[α] ₅₇₈ (room temp.) (CHCl ₃)	U.v. (95% EtOH) λ _{max} /nm (ε)	C.d. (95% EtOH) λ/nm (Δε)
(4a)	-152° (c 0.35)	246 (3.84), 302 (3.38)	243 (-0.747), 277 (+0.173), 312 (-0.180)
(6d)	-67° (c 0.9)	247 (3.84), 279 (3.08), 286 (3.02)	246 (-2.72)
(7) ^a	+24° (c 0.5)	264 (3.74), 306 (3.36)	260 (-0.513), 317 (+0.231)

^a Inseparable mixture of compounds with R = H and Et.

temperature may be expected to influence the results different conditions were also investigated. As is evident from Table 1 (entries 2 and 3) no major variation in e.e. values was found.

It is interesting that replacement of (-)-menthol by (-)-borneol {[α]₅₇₈ (room temp.) -36.7°, c = 4.3, CHCl₃} also induced chirality, albeit to a smaller degree (entry 4). Other imines (3b—d) were also examined. The e.e. value for the propionyl derivative (3b) appeared almost the same in both menthol and borneol (entries 5 and 6) although in menthol the chemical yield was considerably lower. For the phenylindoline (3c) as well as the styryl analogue (3d)† e.e. values of 29 and 22%, respectively, were determined. In view of the low solubility of both compounds the [α]₅₇₈ (room temp.) values were taken after *N*-acetylation (entries 7 and 8).

The present 1,5-electrocyclization results are of great mechanistic interest since formally a simple solvent effect suffices to induce molecular asymmetry.⁸ Moreover to the best of our knowledge the reported cases are the first examples of enantioselective electrocyclic reactions and in principle allow the asymmetric synthesis of indole alkaloids *via* this novel route.⁶ In the latter respect it seemed of general interest to determine the absolute configuration of the indolines (4) by comparison of their c.d. spectra with the corresponding o.r.d. data for the aspidospermidines.⁹ Since the u.v. behaviour of these compounds and of the newly synthesized spiroimides is primarily determined by the aniline chromophore, a comparison between both types of indoline derivatives is justified and may give direct information on the configuration at the



† Satisfactory analytical and spectral data were obtained for all new compounds.

spiro carbon atom. As follows from Table 2 the imide (4a) exhibits a negative Cotton effect at 243 nm which is intensified after *N*-acetylation. In the alkaloid series a similar observation indicates an 11 α -configuration for aspidospermidine.¹⁰ A 2,3-*cis*- α -configuration therefore is assigned to the indoline (4a). Moreover upon reduction of (6c) with LiAlH₄ a mixture of *N*-unsubstituted and *N*-ethyl indolines (7) was obtained¹¹ which also showed analogous c.d. behaviour (Table 2).

We conclude that the use of chiral secondary alcohols as co-solvents in base-catalysed 1,5-electrocyclizations gives rise to enantioselective formation of indolines.

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